Oxazoline N-Oxide Mediated [3+2] Cycloadditions: Application to a Formal Synthesis of a (+)-β-Methylcarbapenem

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[3+2] Cycloaddition between a camphor-derived oxazoline N-oxide $\mathbf{9}$ and the γ , δ -unsaturated enamino ester $\mathbf{11}$ afforded the single adduct $\mathbf{6}$. A stereoselective reduction of the enamino ester side chain allowed the control of the absolute configuration of the two additional asymmetric centres. Nitrogen

protection and oxidative hydrolysis of the resulting product 13, followed by further functional group manipulations, led to the β -lactam derivative 1, a known precursor of the β -methylthienamycin derivative 2a.

Introduction

Oxazoline *N*-oxide mediated cycloadditions have proved to be a valuable tool for the stereoselective synthesis of various natural or unnatural products.^[1] We report here an extension of this methodology to a formal synthesis of a β -methylcarbapenem derivative 1 (Figure 1), synthetic precursor of carbapenem 2a,^[2] which is structurally related to thienamycin 2b.^[3] 1β -Methylcarbapenems display a broadspectrum antibacterial activity, a good chemical stability and a resistance to renal dehydropeptidase DHP1. The major challenge in the synthesis of β -methylcarbapenem antibiotics is the stereocontrolled construction of four contiguous stereogenic centres. Two strategies have been used in previous syntheses of this family of antibiotics: either the

TBDMSO H $\frac{H}{z}$ OAc $\mathbf{2a}: R_1 = Me; R_2 = C(NH)NMe_2$ $\mathbf{2b}: R_1 = H; R_2 = CH_2NH_2$

Figure 1. Thienamycin and β-methylthienamycin derivatives

four stereogenic centres are introduced diastereoselectively, or azetidinone 3, in which three stereogenic centres are already set up, is used as starting material.^[4]

In our retrosynthetic plan, compound **4**, a synthetic precursor of β -methylcarbapenem **1**, could be obtained after oxidative hydrolysis of compound **5**. The two additional stereogenic centres in **5** could be controlled by a diastereoselective reduction of the side chain in the camphor derivative **6** which is the result of a Blaise reaction between adduct **7** and the Reformatsky reagent **8**. According to previous studies, adduct **7** could be obtained through a [3+2] cycloaddition between dipole **9** and 2-butenenitrile **10** (Scheme 1).

Results and Discussion

In relation with other studies directed towards the synthesis of the carbapenem antibiotic carpetimycin A, two routes were examined to prepare compound 6. Thus, as in our initial planning, a cycloaddition reaction between dipole 9 and 2-butenenitrile 10 afforded smoothly the expected adduct 7 as the major product of the reaction. [5] In a more convergent synthesis, a cycloaddition between the γ , δ -unsaturated enamino ester 11, [6] produced by a Blaise condensation [7] between the organozinc reagent 8 and nitrile 10, was also examined. A surprising reactivity was ob-

$$\begin{array}{c} \text{HO H } \stackrel{\text{H}}{\text{H}} \\ \text{ONH} \\ \text{OODE} \\ \text{OOD$$

Scheme 1. β-Methylthienamycin synthetic precursor; retrosynthetic analysis

served in this case and adduct **6** was obtained after only two hours at 40 °C.^[8] Longer reaction times induced the degradation of the adduct. Adduct **6** can be also prepared in a less convergent and less efficient route by Blaise con-

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$$Me \xrightarrow{CN} + CO_2Me \xrightarrow{a, b} Me \xrightarrow{NH_2} CO_2Me$$

$$10 \qquad 11 \qquad Me$$

$$0 \qquad Me \qquad 7 \qquad e, f \ NC$$

$$0 \qquad Me \qquad 0 \qquad Me$$

$$11 \qquad 6 \qquad H_0 \qquad Me$$

$$0 \qquad 0 \qquad Me$$

$$0 \qquad 0 \qquad Me$$

$$0 \qquad 0 \qquad 0 \qquad Me$$

$$0 \qquad 0 \qquad 0 \qquad 0 \qquad Me$$

$$0 \qquad 0 \qquad 0 \qquad 0 \qquad 0 \qquad 0 \qquad 0$$

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Scheme 2. Cycloadditions of oxazoline *N*-oxide with 2-butenenitrile and γ , δ -unsaturated enamino ester: (a) THF, reflux, 1 h; (b) aq. K₂CO₃, 60%; (c) CH₂Cl₂ CaCO₃, MS 4Å, 40 °C, 16 h, 68%; (d) CH₂Cl₂, MS 4Å, 40 °C, 2 h, 54%; (e) BrCHMeCO₂Me, Zn, THF, reflux, 45 min; (f) aq. K₂CO₃, 62%

A da H₂N 33 4 H₂N 134 Me 12b O OMe

HO OME

Scheme 3. Stereoselective reduction of enamino ester adduct: (6)

Scheme 3. Stereoselective reduction of enamino ester adduct: (a) NaBH₃CN, MeOH/CH₂Cl₂, 1:1, MeOH, 2 N, HCl; (b) CbzOSu, 5% aq.NaHCO₃, CH₂Cl₂, 20 °C, 1 h; **13a**: 67%, **13b**: 28%

densation between the nitrile derivative 7 and the Reformatsky reagent 8 (Scheme 2).

The corner stone step of the synthesis was the reduction of the enamino ester side chain in **6**. Sodium cyanoborohydride reduction^[9] of the iminium intermediate **6**, when performed at low temperature, afforded stereoselectively a major product **12a**^[10] along with a minor isomer, **12b** in a 70:30 ratio.^[11] Purification of this mixture of isomers was achieved either by crystallisation in pentane or by chromatography after nitrogen protection as a carbobenzyloxy urethane. Thus, compounds **13a** and **13b** were isolated in 67% and 28% yield, respectively, from adduct **6**.

The modest diastereoselectivity (ca. 40%) observed during the protonation step giving rise to iminium intermediates **6a** or **6b**, could be the result of a noncontrolled conformational equilibrium of the side chain and also of a poor 1,3 asymmetric induction. In contrast, the hydride attack on the iminium intermediates **6a** or **6b**, in which a preferential conformation of the side chain due to hydrogen bonding between the iminium hydrogen and the two oxygens of the isoxazolidino-oxazolidino bicyclic framework is found, could explain the total diastereoselectivity observed in this step despite the configuration of the neighbouring carbon (Scheme 3).

Oxidative hydrolysis, followed by oxidation^[12] of the aldehyde intermediate **14** afforded the expected β-amino acid derivative **15** in 48% overall yield. Ketol **16**, which is a precursor of the chiral auxiliary, is recovered at this stage.^[13] Reductive cleavage of the Cbz group proved to be difficult and an excess of palladium on charcoal was necessary. DCC lactamisation^[14] of the resulting β-amino acid **17**^[15] gave rise to the expected β-lactam derivative **18**. The coupling constants in the ¹H NMR spectrum of β-lactam **18** are in agreement with the literature data^[16] and allowed the determination of the absolute configurations for the asymmetric centres.^[14,17] Under Mitsunobu reaction conditions, the β-lactam **18** afforded the benzoyl derivative **19**. Final selective methanolysis gave rise to the known β-lactam **1**,^[17,18] a pre-

cursor of the β -methylthienamycin derivative **2a** (Scheme 4). The same sequence of reactions performed from the minor compound **13b** afforded a β -lactam epimeric at the methyl side chain. The coupling constants in the 1H NMR spectrum are in agreement with the literature data for this compound. [16,17]

Scheme 4. Oxidative hydrolysis and completion of the β-lactam synthesis: (a) (i) mCPBA, Et_2O , 20 °C, 30 min.(ii) $\text{Na}_2\text{S}_2\text{O}_4$, NaHCO_3 , H_2O ; (b) THF, 2 N HCl, 20 °C, 15 min; (c) NaClO_2 , 10% aq. $\text{NaH}_2\text{PO}_4/\text{tBuOH}$, 1:4, 2-methyl-2-butene, 20 °C, 40 min, 48% from 13a; (d) H_2 , Pd–C, MeOH, 20 °C, 2 h; (e) DCC, MeCN, 60 °C, 3 h, 73% from 15; (f) PhCO $_2\text{H}$, PPh $_3$, DIAD, THF, 2 h30, 20 °C, 77%; (g) MeONa, 10%, MeCN, 0 °C, 3 h. 0.2 N HCl, 67%

In order to clarify the role of the methyl group on the side chain during the reduction of the iminium intermediate, the same reaction was also studied on the desmethyl analogue, a formal precursor of thienamycin **2b** itself. Accordingly, adduct **7** was submitted to the Blaise reaction in the presence of the methyl bromoacetate Reformatsky reagent, to afford the enamino ester **20**. As previously, this compound was also prepared in a more convergent route by a cycloaddition reaction under mild reaction conditions between the oxazoline *N*-oxide **9** and the enamino ester **21**. [19] Low temperature sodium cyanoborohydride reduction of the camphor-derived enamino ester **20** afforded the

expected β -amino ester derivatives **22a** and **22b**. These compounds were directly protected as benzyl carbamates. At this stage, chromatographic purification allowed the isolation of the two isomeric carbamates **23a** and **23b** in 67% and 4% yields, respectively (*de* 89%) (Scheme 5). This result is in good agreement with the diastereoselectivity observed in the β -methylthienamycin series and confirms other observations concerning the diastereoselectivity of this reduction step. [20]

Scheme 5. Cycloaddition of oxazoline N-oxide with an unsubstituted γ , δ -unsaturated enamino ester and stereoselective reduction: (a) CH₂Cl₂, 40 °C, 2 h, 71%; (b) BrCH₂CO₂Me, Zn, THF, reflux, 45 min, K₂CO₃, H₂O, 60%; (c) NaBH₃CN, 1 м in THF, MeOH/2 N HCl, MeOH/CH₂Cl₂, 1:2, -90 °C, 15 h; (d) CbzOSu, NaHCO₃, H₂O, CH₂Cl₂, 20 °C, 1 h, **23a**, 67% from **20**, **23b**, 4% from **20**

Conclusion

In terms of efficiency, this synthesis is in the same range as Fukumoto's synthesis [21] which uses an intramolecular asymmetric nitrone cycloaddition. Besides the preparation of a precursor of β -methylthienamycin, this synthesis also illustrates the possibility of chemio- and stereoselective transformations on adducts such as $\pmb{6}$ in which an alcohol and a carbonyl group are inherently protected. An extension of this methodology to other natural products syntheses is in progress.

Experimental Section

General: ¹H and ¹³C NMR spectra were recorded at 250 MHz and 62.5 MHz, respectively. – Optical rotations were recorded at 20 °C. – Elemental analyses were performed at the CNRS, Gif sur Yvette, France. – Unless otherwise stated, chromatographic purifications were performed on columns with 230–400 mesh silica gel (Merck 9385) and the indicated solvent system. Dichloromethane, acetonitrile and trimethyl orthoacetate were distilled from calcium

hydride. Toluene, diethyl ether and THF were distilled from sodium metal/benzophenone ketyl. Methanol was distilled from magnesium. Zinc powder was activated by successive washings with 2 N hydrochloric acid, water, ethanol and acetone, and was stored under argon in a Schlenk. *trans*-Crotonitrile (10) was prepared isomerically pure from crotonic acid by the method described by Lohaus. [22] Chloroform used for optical measurements was filtered through basic alumina before use. All nonaqueous reactions were performed under an argon atmosphere using oven-dried glassware.

Methyl (2Z,4E)-3-Amino-2-methyl-2,4-hexadienoate (11): The following procedure is typical for Blaise reactions with nitriles. A suspension of activated zinc powder (3.27 g, 50 mmol) in tetrahydrofuran (30 mL) was heated at reflux and a few drops of methyl 2bromopropionate were added. When the reaction mixture turned green, trans-crotonitrile 10 (670 mg, 10 mmol) was added in one portion, stirring was set in motion, and methyl 2-bromopropionate (2.3 mL, 2 mmol) was added dropwise over 45 min. After completion of the addition, the mixture was cooled to room temperature, diluted with tetrahydrofuran (90 mL) and an aqueous 50% potassium carbonate solution (13 mL) was added. The mixture was vigorously stirred for 30 min, then filtered through a pad of celite, which was then rinsed with tetrahydrofuran. The filtrate was concentrated in vacuo and the oily residue taken up with dichloromethane (100 mL), washed with brine (100 mL), dried (Na₂SO₄), filtered and concentrated in vacuo. The crude product was purified by chromatography on florisil (100-200 mesh, 50% ether/pentane, $R_{\rm f} = 0.8$) to give the title compound as a yellow oil (930 mg, 63%). A lower yield was obtained when purification was performed on silica gel. – IR (film): $\tilde{v} = 3400$, 1730, 1650 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): $\delta = 6.35$ (d, J = 16.2 Hz, 1 H, C4-H), 6.20 (m, 3 H, C5-H and NH₂), 3.65 (s, 3 H, OMe), 1.85 (dd, J = 6.3 Hz, 1 Hz, 3 H, C6-H), 1.80 (s, 3 H, CH₃-C2). - ¹³C NMR (62.5 MHz, CDCl₃): $\delta = 171.5$ (CO), 153.1 (C3), 130.7 (C4), 127.1 (C5), 90.9 (C2), 50.7 (OMe), 18.5 (CH₃-C2), 12.1 (C6). - MS (CI, NH₃); m/z: 156 [MH⁺].

Methyl (2S,3S,3aS,4aS,5R,8S,8aR,2'Z)-3-(2,5,10,10-Tetramethyloctahydro-2H-5,8-methano-isoxazolo[3,2-b]benzoxazol-3-yl)-3'amino-2'-methyl-2'-propenoate (6). - Cycloaddition Between Oxazoline N-Oxide (9) and Enamino Ester 11: This procedure is typical for cycloadditions between oxazoline N-oxide (9) and conjugated enamino esters. To a suspension of (+)-3-hydroxyaminoisoborneol hydrochloride (4 g, 18 mmol) in dichloromethane was added trimethyl orthoformate (7.9 mL, 72 mmol, 4 equiv.). The white suspension was stirred at 40 °C for 3.5 h. Powdered 4-A molecular sieves (2 g) were then added and the mixture stirred for an additional 30 min. The enamino ester 11 (8.4 g, 54 mmol, 3 equiv.) was then added and the reaction mixture stirred for 2 h at reflux. After cooling to room temperature, the solution was filtered through a pad of celite, which was then rinsed with dichloromethane. The filtrate was concentrated in vacuo and the residue purified by chromatography (50% ether/pentane, $R_f = 0.5$) to give the cycloadduct 6 as a white solid (3.4 g, 54%). – IR (film moistened with CHCl3): \tilde{v} = 3400, 1730 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): $\delta = 6.70$, (s, 2 H, NH₂), 5.43 (d, J = 5.6 Hz, 1 H, C3a-H), 4.10 (dq, J = 10.8 Hz, J = 5.7 Hz, 1 H, C2-H, 3.91 (d, <math>J = 7.6 Hz, 1 H, C4a-H), 3.70 (s, J)3 H, OMe), 3.55 (d, J = 7.6 Hz, 1 H, C8a-H), 3.05 (dd, J = 10.8, J = 5.6 Hz, 1 H, C3-H), 2.12 (d, J = 4 Hz, 1 H, C8-H), 1.80 (s, 3 H, Me-C2'), 1.70 (m, 2 H, C7-H × 2), 1.40 (m, 1 H, one of C6-H), 1.25 (d, J = 6 Hz, 3 H, Me-C2), 1.15 (m, 1 H, one of C6-H), 0.95, 0.90, 0.80 (3s, 9 H, $CH_3 \times 3$). – ¹³C NMR (62.5 MHz, CDCl₃): $\delta = 171.2$ (CO), 153.9 (C3'), 99.3 (C3a), 92.4 (C2'), 90.7 (C4a), 78.4 (C2), 75.3 (C8a), 55.0 (C3), 50.7 (OMe), 49.6 (C8), 49.1

(C5), 45.0 (C10), 31.6 (C6), 25.8 (C7), 15.8 (Me–C2'), 12.4 (Me–C2), 22.3, 18.6, 10.9 (CH₃ × 3). – MS (CI, NH₃); m/z: 351 [MH⁺], 180, 172, 156. – [α]²⁰₂₀ = –161 (c = 1, CHCl₃). – C₁₉H₃₀N₂O₄ (350.46195): calcd. C 65.12, H 8.63, N 7.99; found C 64.72, H 8.62, N 7.59.

(2S,3S,3aS,4aS,5R,8S,8aR)-3-(2,5,10,10-Tetramethyl-octahydro-2*H*-5,8-methano-isoxazolo[3,2-*b*]benzoxazol-3-carbonitrile (7). Cycloaddition Between Oxazoline N-Oxide (9) and Crotonitrile (10): To a suspension of 3-hydroxylaminoisoborneol hydrochloride (8 g, 36.mmol) and calcium carbonate (3.6 g, 36 mmol, 1 equiv.) in dichloromethane (150 mL) was added trimethyl orthoformate (15.8 mL, 145 mmol, 4 equiv.). The white suspension was stirred for 4 h at reflux, then trans-crotonitrile 10 (20.8 g, 300 mmol, 8 equiv.), was added. The mixture was stirred overnight at reflux, then cooled to room temperature and filtered through a pad of celite, eluting with dichloromethane. The filtrate was concentrated in vacuo and the residue purified by chromatography (15% ethyl acetate/heptane) to give, in order of elution, a small amount of the exo cycloadduct (10%), followed by the endo cycloadduct 7 (6.6 g, 70%), which was isolated as a white powder. - IR (film moistened with CHCl₃): $\tilde{v} = 2950$, 2260 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): $\delta = 5.43$ (d, J = 6.7 Hz, 1 H, C3a-H), 4.28 (dq, J = 10.1, 5.9 Hz, 1 H, C2-H), 4.12 (d, J = 7.5 Hz, 1 H, C4a-H), 3.31 (d, J = 7.5 Hz, 1 H, C8a-H), 2.92 (dd, J = 10.1, 6.7 Hz, 1 H, C3-H), 2.12 (d, J =4 Hz, 1 H, C8-H), 1.70 (m, 2 H, C7-H \times 2), 1.42 (d, J = 6 Hz, 3 H, Me-C2), 1.40 (m, 2 H, C6-H × 2), 0.96, 0.88, 0.78 (3s, 9 H, $CH_3 \times 3$). – ¹³C NMR (62.5 MHz, CDCl₃): $\delta = 115.2$ (CN), 97.0 (C3a), 90.2 (C4a), 75.6 (C2), 75.1 (C8a), 48.9 (C3), 48.6 (C5), 45.8 (C10), 45.5 (C8), 31.3 (C6), 25.5 (C7), 16.5 (Me-C2), 22.1, 18.7, 10.6 (CH₃ × 3). – MS (CI, NH₃); m/z: 263 [MH⁺], 180. – $[\alpha]_D^{20}$ = -167 (c = 1, CHCl₃). $-C_{15}H_{22}N_2O_2$ (262.35497): calcd. C 68.67, H 8.45, N 10.68; found C 68.22, H 7.99, N 10.25.

Preparation of Compound 6 by Blaise Reaction of Cycloadduct 7: This reaction was carried out according to the typical procedure for Blaise reactions described for the preparation of 11, starting from cycloadduct 7 (260 mg, 1 mmol), zinc powder (327 mg, 5 mmol), and methyl 2-bromopropionate (450 μ L, 4 mmol). Purification of the crude product by chromatography (40% pentane/ ether, $R_{\rm f} = 0.55$) gave compound 6 as a white solid (217 mg, 62%), identical in all respect to the one prepared by cycloaddition with enamino ester 11.

(2S,2'RS,3S,3'S,3aS,4aS,5R,8S,8aR)-3-(2,5,10,10-Tetramethyl-octahydro-2H-5,8-methano-isoxazolo[3,2-b]benzoxazol-3-yl)-3'-amino-2'-methylpropanoate (12): A solution of the enamino ester 6 (3.33 g, 9.5 mmol) in 2:1 dichloromethane/methanol (93 mL) containing a few crystals of bromocresol green was cooled to -90 °C with stirring and a few drops of a 2 N solution of hydrogen chloride in methanol (prepared by careful addition of acetyl chloride to anhydrous methanol at 0 °C) were added until the solution colour turned from blue to yellow. Sodium cyanoborohydride (779 mg, 12.5 mmol, 1.3 equiv.) was then added in one portion, resulting in a blue colour. A 2 N hydrogen chloride solution in methanol (30 mL) was then added dropwise over 15 h (syringe pump), in order to maintain a persistent yellow colour. After completion of the addition, the solution was slowly warmed to -20 °C, and a 0.1 N sodium hydroxide solution was added until the solution turned deep blue. Brine (50 mL) was added and the mixture was extracted with dichloromethane (2 × 100 mL). The combined organic layer was dried (Na₂SO₄), filtered and concentrated in vacuo to give a mixture of isomeric β-amino esters 12a and 12b (3.3 g; 99%), which were not separated at this stage but immediately carried forward to the next reaction.

Methyl (2S,2'R,3S,3'S,3aS,4aS,5R,8S,8aR)-3-(2,5,10,10-Tetramethyl-octahydro-2*H*-5,8-methano-isoxazolo[3,2-*b*]benzoxazol-3-yl)-3'-(phenylmethyloxycarbonyl)amino-2'-methylpropanoate (Benzyloxycarbonyloxy)succinimide (CbzOSu, 2.8 g, 11.25 mmol, 1.2 equiv.) was added to a solution of amines 12a and 12b (3.3 g, 9.4 mmol) in a mixture of dichloromethane (100 mL) and aqueous 5% sodium hydrogen carbonate solution (100 mL). The biphasic mixture was vigorously stirred for 1 h and the layers separated. The aqueous layer was extracted with dichloromethane (100 mL) and the combined organic layer was dried (Na₂SO₄), filtered and concentrated in vacuo. Purification of the crude product by chromatography (30% ethyl acetate/pentane) gave, in order of elution, the major isomer 13a ($R_f = 0.65, 3.07 \text{ g}, 67\%$), followed by the minor isomer **13b** ($R_f = 0.4$, 1.28 g, 28%). – **13a:** IR (CHCl₃): $\tilde{v} = 3400$, 1740, 1680 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): $\delta = 7.40$ (broad s, 5 H, Ar-H), 5.35 (d, J = 5.6 Hz, 1 H, C3a-H), 5.30 (d, J =7.5 Hz, 1 H, NH), 5.15 (m, 2 H, Ar–CH₂), 4.15 (dq, J = 10.6 Hz, J = 6.4 Hz, 1 H, C2-H, 3.70 (s, 3 H, OMe), 3.69 (obscured m, 2)H, C4a-H and C3'-H), 3.35 (d, J = 7.5 Hz, 1 H, C8a-H), 2.80 (m, 1 H, C2'-H), 2.10 (d, J = 4.2 Hz, 1 H, C8-H), 1.80 (m, 1 H, C3-H), 1.70 (m, 2 H, C7-H \times 2), 1.40 (m, 2 H, C6-H), 1.20 (2d, J =6.4 Hz, 6 H, Me-C2 and Me-C2'), 0.95, 0.85 and 0.80 (3s, 9 H, $CH_3 \times 3$). – ¹³C NMR (62.5 MHz, CDCl₃): $\delta = 174.1$ (CO₂), 155.7 (NHCO₂), 136.2, 127.9, 127.5 (Ar), 99.2 (C3a), 89.5 (C4a), 77.9 (C2), 72.8 (C8a), 66.1 (CH₂-Ar), 54.6 (C3'), 51.4 (OMe), 50.5 (C3), 49.0 (C8), 48.3 (C5), 44.7 (C10), 43.1 (C2'), 31.1 (C6), 25.1 (C7), 15.4 (Me-C2), 14.4 (Me-C2'), 21.7. 18.2 and 10.4 (CH₃ \times 3). – MS (CI, NH₃); m/z: 487 [MH⁺], 308, 180. – $[\alpha]_D^{20} = -57.3$ (c = 1, CHCl₃). - HRMS calculated for C₂₇H₃₈N₂O₆: 487.28081; found 487.28086. – **13b:** IR (CHCl₃): $\tilde{v} = 3400$, 1740, 1680 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): $\delta = 7.40$ (broad s, 5 H, Ar-H), 5.40 (d, J = 10.4 Hz, 1 H, NH), 5.30 (d, J = 6.1 Hz, 1 H, C3a-H), 5.10(m, 2 H, Ar-CH₂), 4.15 (dq, J = 10.6 Hz, J = 6.4 Hz, 1 H, C2-H), 3.70 (obscured m, 2 H, C4a-H and C3'-H), 3.65 (s, 3 H, OMe), 3.30 (d, J = 7.5 Hz, 1 H, C8a-H), 2.80 (m, 1 H, C2'-H), 2.10 (d,J = 4.2 Hz, 1 H, C8-H), 1.90 (m, 1 H, C3-H), 1.70 (m, 2 H, C7-H) $H \times 2$), 1.40 (m, 2 H, C6-H), 1.30 (d, J = 5.8 Hz, 3 H, Me-C2'), 1.25 (d, J = 7 Hz, 3 H, Me-C2), 0.95, 0.85 and 0.80 (3s, 9 H, $CH_3 \times 3$). – ¹³C NMR (62.5 MHz, CDCl₃): $\delta = 174.5$ (CO₂), 155.9 (NHCO₂), 136.7, 128.4, 128.0 (Ar), 99.4 (C3a), 90.1 (C4a), 78.0 (C2), 73.5 (C8a), 66.5 (CH₂-Ar), 54.9 (C3'), 51.8 (OMe), 51.0 (C3), 49.4 (C8), 48.7 (C5), 45.3 (C10), 43.6 (C2'), 31.5 (C6), 25.5 (C7), 16.1 (Me-C2), 15.2 (Me-C2'), 22.1, 18.6 and 10.9 (CH₃ \times 3). – MS (CI, NH₃); m/z: 487 [MH⁺], 308, 180. – $[\alpha]_D^{25} = -56.3$ (c = 1, CHCl₃). – HRMS calculated for $C_{27}H_{38}N_2O_6$: 487.28081; found 487.28081.

Methyl (2R,3S,4S,5S) 4-Formyl-5-hydroxy-2-methyl-3-(phenylmethoxycarbonyl)aminohexanoate (14): m-Chloroperoxybenzoic acid (1.22 g, 7.1 mmol, 5 equiv.) was added in one portion to a solution of amine 13a (691 mg, 1.42 mmol) in diethyl ether (35 mL). The clear solution was stirred for 30 min at room temperature and the unchanged peracid was quenched by addition of an aqueous 10% sodium hydrogen carbonate/10% sodium thiosulfate (35 mL). The biphasic solution was vigorously stirred at room temperature for 30 min, then the layers were separated. The organic layer was washed with water, brine (25 mL each), dried (Na₂SO₄), filtered and concentrated in vacuo to give a white foam which was immediately redissolved in tetrahydrofuran (35 mL). A 2 N hydrochloric acid solution (35 mL) was added and the mixture was stirred at room temperature for 15 min, then extracted with ethyl acetate (3 × 50 mL). The combined organic extracts were washed successively with 5% sodium hydrogen carbonate solution, pH 7 buffer solution and brine (50 mL each), dried (Na₂SO₄) and then

filtered and concentrated in vacuo. Purification of the crude product by chromatography (50% ethyl acetate/pentane) gave, in order of elution, the recovered ketol 16 ($R_{\rm f} = 0.65, 210 \, {\rm mg}, 88\%$ recovery), followed by the aldehyde 14 ($R_f = 0.4$, 335 mg, 70% for two steps) as an amorphous white solid. – IR (CHCl₃): $\tilde{v} = 3400$, 1740, 1720 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): δ = 9.90 (broad s, 1 H, CHO), 7.30 (broad s, 5 H, Ar-H), 5.60 (d, J = 10.3 Hz, 1 H, NH), 5.10 (m, 2 H, Ar-CH₂), 4.20 (m, 2 H, C3-H and C5-H), 3.70 (s, 3 H, OMe), 2.80 (m, 1 H, C4-H), 2.60 (m, 1 H, C2-H), 1.40 (d, J =6.3 Hz, 3 H, C6-H \times 3), 1.20 (d, J = 7.1 Hz, 3 H, Me-C2). - ¹³C NMR (62.5 MHz, CDCl₃): $\delta = 205.7$ (CHO), 174.8 (CO₂Me), 156.4 (NHCO₂), 136.1, 128.4, 128.0, 127.8 (Ar), 67.5 (C5), 66.8 (Ar-CH₂), 58.9 (C3), 52.0 (C4), 51.9 (OMe), 42.9 (C2), 22.2 (C6), 13.8 (Me-C2). – MS (CI, NH₃); m/z: 338 [MH⁺], 320, 294, 160. – $[\alpha]_{D}^{20} = +46 \ (c = 0.83, \text{ CHCl}_3). - C_{17}H_{23}NO_6 \ (337.37596)$: calcd. C 60.52, H 6.87, N 4.15; found C 60.41, H 6.91, N 4.10.

Methyl (2R,3S,4S,5S)-4-Carboxy-5-hydroxy-2-methyl-3-(phenylmethoxycarbonyl)aminohexanoate (15): To a solution of the aldehyde 14 (200 mg, 0.6 mmol) and 2-methyl-2-butene (7.5 mL) in tert-butyl alcohol (20 mL) was added an aqueous 10% sodium chlorite/10% sodium dihydrogen phosphate solution (7 mL). The colourless solution was stirred at room temperature for 30 min. Water (40 mL) was added and the mixture was extracted with ethyl acetate $(5 \times 50 \text{ mL})$. The combined organic extracts were washed with brine (100 mL), dried (Na₂SO₄), filtered and concentrated in vacuo. The crude product was purified by preparative thin-layer chromatography (10% methanol/dichloromethane). The baseline was collected to give the acid 15 as a white solid (145 mg, 69%). - IR (CHCl₃): $\tilde{v} = 3500$, 3400, 1760, 1740 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): $\delta = 7.30$ (broad s, 5 H, Ar-H), 5.90 (d, J = 10.3 Hz, 1 H, NH), 5.10 (m, 2 H, Ar-CH₂), 4.20 (m, 1 H, C5-H), 4.10 (m, 1 H, C3-H), 3.65 (s, 3 H, OMe), 2.65 (m, 1 H, C4-H), 2.50 (m, 1 H, C2-H), 1.40 (d, J = 6.2 Hz, 3 H, C6-H \times 3), 1.20 (d, J = 7 Hz, 3 H, Me-C2). – 13 C NMR (62.5 MHz, CDCl₃): $\delta = 178.0$ (CO₂H), 175.1 (CO₂Me), 156.6 (NHCO₂), 136.3, 128.4, 128.1, 127.9 (Ar), 67.6 (C5), 66.9 (Ar-CH₂), 55.1 (C3), 52.0 (C4), 51.7 (OMe), 43.7 (C2), 21.4 (C6), 14.1 (Me-C2). – MS (CI, NH₃); m/z: 354 [MH⁺], 310. $- [\alpha]_D^{20} = -1.5$ (c = 0.74, CHCl₃). - HRMS calculated for C₁₇H₂₃NO₇: 376.13722; found 376.13697.

Methyl (2*R*,3*S*,4*S*,5*S*)-3-Amino-4-carboxy-5-hydroxy-2-methylhexanoate (17): To a solution of the acid 15 (145 mg, 041 mmol) in anhydrous methanol (13 mL) was added 10% palladium on carbon (145 mg) and the mixture was stirred under an atmosphere of hydrogen for 2.5 h. The black suspension was then filtered through a pad of celite, eluting with methanol. The filtrate was concentrated in vacuo to give the unprotected amino acid 17 as a white foam (90 mg, 99%) which was used in the next reaction without further purification. – IR (CHCl₃): \tilde{v} = 3500, 3400, 1760, 1740 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): δ = 4.00 (m, 1 H, C5-H), 4.10 (m, 1 H, C3-H), 3.70 (s, 3 H, OMe), 2.90 (m, 1 H, C4-H), 2.30 (m, 1 H, C2-H), 1.40 (d, 6 H, 3 × C-H and Me–C2). – ¹³C NMR (62.5 MHz, CDCl₃): δ = 177.4 (CO₂H), 175.2 (CO₂Me), 68.1 (C5), 53.2 (C3), 52.1 (OMe), 50.0 (C4), 41.7 (C2), 22.2 (C6), 12.8 (Me–C2). – MS (CI, NH₃); m/z: 220 [MH⁺], 144.

(3S,4S)-3-[(1S)-Hydroxyethyl]-4-[(1R)-methoxycarbonyl]ethyl-2-azetidinone (18): A solution of the amino acid 17 (90 mg, 0.41 mmol) and dicyclohexylcarbodiimide (170 mg, 0.82 mmol, 2 equiv.) in acetonitrile (5 mL) was stirred at reflux for 1.5 h. After cooling to room temperature, the solution was diluted with ethyl acetate (5 mL) and the solids were removed by filtration. The filtrate was concentrated in vacuo and the residue purified by preparative thin-layer chromatography (10% methanol/dichloromethane,

 $R_{\rm f}=0.4$) to give the β-lactam **18** as a colourless oil (61 mg, 73%). – IR (film): $\tilde{\rm v}=3400,\ 1740,\ 1735\ {\rm cm^{-1}}.\ -\ ^1{\rm H}\ {\rm NMR}\ (250\ {\rm MHz},\ {\rm CDCl_3})$: $\delta=6.30$ (s, 1 H, NH), 4.11 (m, 1 H, C*H*–OH), 3.71 (s, 3 H, OMe), 3.69 (d, $J=2.2\ {\rm Hz},\ 1$ H, C4-H), 3.06 (dd, $J=6.3\ {\rm Hz},\ J=2.2\ {\rm Hz},\ 1$ H, C3-H), 2.67 (m, 1 H, C*H*–CO₂Me), 1.35 (d, $J=6.3\ {\rm Hz},\ 3$ H, C*H*₃–CHOH), 1.28 (d, $J=7.1\ {\rm Hz},\ 3$ H, C*H*₃–CH–CO₂Me). – ¹³C NMR (62.5 MHz, CDCl₃): $\delta=174.3\ ({\rm CO_2Me}),\ 168.6\ ({\rm C2}),\ 65.8\ ({\rm CHOH}),\ 61.4\ ({\rm C4}),\ 52.6\ ({\rm C3}),\ 52.1\ ({\rm OMe}),\ 43.0\ ({\rm CHCO_2Me}),\ 21.0\ ({\rm CH_3}$ –CHOH), 13.4 (CH₃–CHCO₂Me). – MS (CI, NH₃); m/z: 219 [MNH₄+], 202 [MH+], 141. – [α]_D²⁰ = -12.5 (c=1, CHCl₃). – C₉H₁₅NO₄ (201.2242): calcd. C 53.72, H 7.51, N 6.96; found C 53.69, H 7.63, N 6.51.

 $(3S,\!4S)\text{-}3\text{-}[(1R)\text{-}Benzoyloxyethyl]\text{-}4\text{-}[(1R)\text{-}methoxycarbonyl]\text{ethyl-}2\text{-}$ azetidinone (19): Benzoic acid (146 mg, 1.2 mmol, 4 equiv.) was added to a cooled (0 °C) solution of azetidinone 18 (60 mg, 0.3 mmol) and triphenylphosphine (274 mg, 1.05 mmol, 3.5 equiv.) in tetrahydrofuran (2 mL). The solution was stirred for 5 min at 0 °C and diisopropylazodicarboxylate (DIAD, 235 µL, 1.2 mmol, 4 equiv.) was added. The cooling bath was removed and the solution stirred at room temperature for 2.5 h. The solvent was evaporated and the residue purified by chromatography (50% ethyl acetate/pentane, $R_{\rm f} = 0.35$) to give the ester 19 as a colourless oil (70 mg, 77%). – IR (film): $\tilde{\nu} = 3300$, 1770, 1735, 1720 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): $\delta = 8.02$ (m, 2 H, Ar-H), 7.57 (m, 1 H, Ar-H), 7.44 (m, 2 H, Ar-H), 6.16 (s, 1 H, NH), 5.50 (dq, J = 7.3 Hz, J = 6.4 Hz, 1 H, CH-OCOPh), 3.90 (dd, J = 6.1 Hz, J = 2.3 Hz, 1 H, C4-H), 3.70 (s, 3 H, OMe), 3.33 (dd, J = 7.3 Hz, J = 2.3 Hz, 1 H, C3-H), 2.73 (dq, J = 7.1 Hz, J = 6.1 Hz, 1 H, CHCO₂Me), 1.48 (d, J =6.4 Hz, 3 H, CH₃-CHOCOPh), 1.24 (d, 3 H, CH₃-CHCO₂Me). – ¹³C NMR (62.5 MHz, CDCl₃): $\delta = 174.4$ (CO₂Me), 166.7 (C2), 165.5 (PhCO₂), 133.1, 130.0, 129.6, 128.3 (Ar), 68.6 (CHOCOPh), 59.2 (C4), 53.7 (C3), 51.9 (OMe), 42.7 (CHCO₂Me), 18.4 (CH₃CHOCOPh), 13.0 (CH₃CHCO₂Me). – MS (CI, NH₃); m/z: $306 \, [MH^+]$. $- [\alpha]_D^{20} = -26.4 \, (c = 0.53, CHCl_3)$. - HRMS calculated for C₁₆H₁₉NO₅: 328.1160; found 328.1163.

(3S,4S)-3-[(1R)-Hydroxyethyl]-4-[(1R)-methoxycarbonyl]ethyl-2azetidinone (1): A solution of the ester 19 (26 mg, 0.085 mmol) in acetonitrile (1 mL) was cooled to 0 °C and a 10% solution of sodium methoxide in methanol (29 μ L) was added. The solution was stirred at 0 °C for 3 h, ethyl acetate (5 mL) was added and the mixture was neutralised by addition of a few drops of 0.2 N hydrochloric acid. The organic layer was washed with brine (10 mL), dried (Na₂SO₄), filtered and concentrated in vacuo. The residue was purified by preparative thin-layer chromatography (10% methanol/ethyl acetate, $R_{\rm f} = 0.64$) to give the azetidinone 1 as a white solid (11.5 mg, 67%), m.p. 102 °C. – IR (film): $\tilde{v} = 3400$, 1770, 1735 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): $\delta = 6.15$ (s, 1 H, NH), 4.16 (m, 1 H, CH–OH), 3.77 (d, J = 7.8 Hz, J = 2.1 Hz, 1 H, C4– H), 3.72 (s, 3 H, OMe), 2.98 (dd, J = 7.0 Hz, J = 2.1 Hz, 1 H, C3-H), 2.67 (m, 2 H, CH-CO₂Me and OH), 1.31 (d, J = 6.3 Hz, 3 H, CH_3 -CHOH), 1.27 (d, J = 7.1 Hz, 3 H, CH_3 -CH-CO₂Me). – ¹³C NMR (62.5 MHz, CDCl₃): $\delta = 174.8$ (CO₂Me), 167.7 (C2), 65.8 (CHOH), 62.6 (C4), 53.6 (C3), 52.2 (OMe), 43.20 (CHCO₂Me), 21.1 (CH₃-CHOH), 13.7 (CH₃-CHCO₂Me). - MS (CI, NH₃); m/z: 219 [MNH₄⁺], 202 [MH⁺]. – [α]_D²⁰ = –44.8 (c = 0.22, CHCl₃). – HRMS (electrospray) calculated for C₉H₁₅NNaO₄ (M + Na): 224.0898; found 224.0896.

Methyl (2Z,4E)-3-Amino-2,4-hexadienoate (21): This reaction was carried out according to the typical procedure for Blaise reactions described for the preparation of 11, starting from crotonitrile 10 (2 g, 30 mmol), zinc powder (9.7 g, 150 mmol) and methyl bromacetate (11.3 mL, 120 mmol). Purification of the crude product by

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chromatography (50% ether/pentane, $R_{\rm f}=0.7$) gave the enamino ester **21** as a white solid (2.8 g, 67%). – IR (film moistened with CHCl₃): $\tilde{v}=3400$, 1730, 1650 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): $\delta=6.25$ (dq, J=16 Hz, J=6.7 Hz, 1 H, C5-H), 6.20 (dd, J=16 Hz, J=1.1 Hz, 3 H, C4-H), 4.65 (s, 1 H, C2-H), 3.65 (s, 3 H, OMe), 1.85 (dd, J=6.7 Hz, J=1.1 Hz, 3 H, C6-H × 3). – ¹³C NMR (62.5 MHz, CDCl₃): $\delta=170.8$ (CO₂Me), 156.1 (C3), 130.6 (C4), 128.3 (C3), 85.5 (C2), 50.3 (OMe), 18.2 (C6). – MS (CI, NH₃); m/z: 142 [MH⁺]. – HRMS (electrospray) calculated for C₇H₁₁NNaO₂ (M + Na): 164.0687; found 164.0687.

Methyl (2S,3S,3aS,4aS,5R,8S,8aR,2'Z)-3-(2,5,10,10-Tetramethyloctahydro-2H-5,8-methano-isoxazolo[3,2-b]benzoxazol-3-yl)-3'amino-2'-propenoate (20): This compound was prepared by cycloaddition between oxazoline N-oxide (9) and enamino ester 21, following the typical procedure described for the preparation of cycloadduct 11, starting from (+)-3-hydroxyaminoisoborneol hydrochloride (1.1 g, 5 mmol), trimethyl orthoformate (2.16 mL, 20 mmol, 4 equiv.) and enamino ester 21 (2.8 g, 20 mmol, 4 equiv.) in the presence of 4Å molecular sieves (500 mg). Purification by chromatography (50% ether/pentane, $R_{\rm f}=0.3$) gave the title compound as a white solid (1.18 g, 71%). - IR (film moistened with CHCl₃): $\tilde{v} = 3400$, 1730 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): $\delta =$ 5.43 (d, J = 7.5 Hz, 1 H, C3a-H), 4.55 (s, 1 H, C2'-H), 4.05 (dq,J = 10.7 Hz, J = 5.6 Hz, 1 H, C2-H), 3.91 (d, J = 7.5 Hz, 1 H,C4a-H), 3.66 (s, 3 H, OMe), 3.48 (d, J = 7.5 Hz, 1 H, C8a-H), 2.40 (dd, J = 10.7 Hz, J = 5.7 Hz, 1 H, C3-H), 2.12 (d, J = 3.8 Hz, 1 H, C8-H), 1.70 (m, 2 H, C7-H), 1.40 (m, 2 H, C6-H), 1.25 (d, J =5.6 Hz, 2 H, Me–C2), 0.95, 0.90, 0.80 (3s, 9 H, $CH_3 \times 3$). – ^{13}C NMR (62.5 MHz, CDCl₃): $\delta = 170.3$ (CO₂Me), 157.7 (C3'), 99.7 (C3a), 90.7 (C4a), 86.1 (C2'), 78.2 (C2), 74.8 (C8a), 59.6 (C3), 50.4 (OMe), 49.6 (C8), 49.1 (C5), 45.4 (C10), 31.7 (C6), 25.8 (C7), 16.3 (Me-C2), 22.4, 18.7, 10.3 (CH₃ \times 3). – MS (CI, NH₃); m/z: 337 $[MH^+]$, 181. – $[\alpha]_D^{20} = -199$ (c = 0.94, CHCl₃). – HRMS (electrospray) calculated for $C_{18}H_{28}N_2NaO_4$ (M + Na): 359.1946; found 359.1956.

Methyl (2S,3S,3'S,3aS,4aS,5R,8S,8aR)-3-(2,5,10,10-Tetramethyloctahydro-2H-5,8-methano-isoxazolo[3,2-b]benzoxazol-3-yl)-3'aminopropanoate (22): A blue solution of the enamino ester 20 (700 mg, 2.08 mmol) in 2:1 dichloromethane/methanol (20 mL) containing few crystals of bromocresol green was cooled to -90 °C with stirring, and few drops of a 2 N methanolic hydrogen chloride solution were added, resulting in a change of colour from blue to yellow. A 1 M sodium cyanoborohydride solution in tetrahydrofuran (3.12 mL, 31.2 mmol, 1.5 equiv.) was added dropwise. When the solution colour turned to blue, a 2 N methanolic hydrogen chloride solution (6.6 mL) was added dropwise over 15 h (syringe pump). After completion of the addition, the yellow solution was progressively warmed to -20 °C and was neutralised by addition of an aqueous 0.1 N sodium hydroxide solution until the colour turned persistent blue. Brine (30 mL) was added, and the mixture extracted by dichloromethane $(3 \times 50 \text{ mL})$. The combined organic extracts were dried (Na₂SO₄), filtered and concentrated in vacuo to give a mixture of isomeric β-amino esters 22a and 22b (700 mg, 99%), which was used in the next reaction without further purification.

Methyl (2*S*,3*S*,3′*S*,3a*S*,4a*S*,5*R*,8*S*,8a*R*)-(2,5,10,10-Tetramethyl-5,8-methano-octahydro-2*H*-isoxazolo[3,2-*b*]benzoxazole-3)-3′-yl-3′-(phenylmethyloxycarbonyl)aminopropanoate (23a): (Benzyloxycarbonyloxy)succinimide (618 mg, 2.5 mmol, 1.2 equiv.) was added to a solution of amines 22a and 22b (700 mg, 2.07 mmol) in a mixture of dichloromethane (20 mL) and aqueous 5% sodium hydrogen carbonate solution (20 mL). The biphasic mixture was vigorously stirred for 1 h, and the layers were separated. The aqueous layer

was extracted with dichloromethane (50 mL) and the combined organic layer was dried (Na₂SO₄), filtered and concentrated in vacuo. Purification of the crude product by chromatography (35% ethyl acetate/pentane) gave, in order of elution, the major isomer 23a $(R_{\rm f} = 0.5, 659 \, {\rm mg}, 67\%)$, followed by the minor isomer 23b $(R_{\rm f} =$ 0.4, 37 mg, 4%). – **23a:** IR (film moistened with CHCl₃): $\tilde{v} = 3400$, 1740 cm⁻¹. – ¹H NMR (250 MHz, CDCl₃): $\delta = 7.25$ (broad s, 5 H, Ar-H), 5.25 (m, 2 H, C3-H and NH), 5.05 (m, 2 H, Ar-CH₂), 4.35 (m, 1 H, C2-H), 3.65 (d, J = 7.4 Hz, 1 H, C4-H), 3.60 (s, 3 H, OMe), 3.35 (d, J = 7.4 Hz, 1 H, C8a-H), 2.60 (m, 2 H, C2'-H \times 2), 2.00 (d, J = 3.8 Hz, 1 H, C8-H), 1.85 (m, 1 H, C3-H), 1.60 (m, 2 m)H, C7-H \times 2), 1.40 (m, 2 H, C6-H \times 2), 1.20 (d, J = 5.6 Hz, 3 H, Me-C2), 0.85, 0.80, 0.70 (3s, 9 H, $CH_3 \times 3$). – ¹³C NMR (62.5 MHz, CDCl₃): $\delta = 170.3$ (CO₂Me), 155.4 (NHCO₂), 136.4, 128.2, 127.8 (Ar), 99.5 (C3a), 89.8 (C4a), 78.0 (C2), 72.9 (C8a), 66.3 (Ar-CH₂), 56.3 (C3'), 51.6 (OMe), 49.2 (C3), 48.5 (C5), 45.3 (C8), 45.0 (C10), 38.6 (C2'), 31.3 (C6), 25.3 (C7), 15.8 (Me-C2), 21.9, 18.4, 10.6 (CH₃ \times 3). – MS (CI, NH₃); m/z: 473 [MH⁺]. – $[\alpha]_{\rm D}^{20} = -53$ (c = 2.1, CHCl₃). – HRMS (electrospray) calculated for $C_{26}H_{36}N_2NaO_4$ (M + Na): 495.24710; found 495.24711.

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- ^[19] Enamino ester **21** was prepared in 67% yield by treatment of (*E*)-2-butenenitrile with ethyl bromoacetate (4 equiv.) and zinc (5 equiv.) in refluxing THF, followed by work up with an aqueous solution of K_2CO_3 .
- the diastereomer corresponding to compound **22b** is the only product of the reaction. This result is also in agreement with a conformational restriction of the side chain due to hydrogen bonding in the iminium intermediate and a diastereoselective attack of hydride at the *Re* face of this iminium (unpublished work).
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